

MATERIAL SAFETY DATA SHEET

1. SUBSTANCE AND SOURCE IDENTIFICATION

National Institute of Standards and Technology
Standard Reference Materials Program
100 Bureau Drive, Stop 2320
Gaithersburg, Maryland 20899-2320

SRM Number: 3148a
MSDS Number: 3148a
SRM Name: Scandium Standard Solution

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Description: This Standard Reference Material (SRM) is intended for use as a primary calibration standard for the quantitative determination of scandium. One unit of SRM 3148a consists of five 10 mL sealed borosilicate glass ampoules of an acidified aqueous solution prepared gravimetrically to contain a known mass fraction of scandium. The solution contains nitric acid at a volume fraction of approximately 10 %.

Material Name: Scandium Standard Solution

Other Designations:

Scandium: Sc; elemental scandium; ekaboron.

Scandium Nitrate: Scandium trinitrate; nitric acid, scandium (3+) salt; scandium (3+) nitrate.

Nitric Acid: Aqua fortis; hydronitrate; azotic acid; engraver's acid.

2. COMPOSITION AND INFORMATION ON HAZARDOUS INGREDIENTS

Component	CAS Registry	EC Number (EINECS)	Concentration (%)
Nitric Acid	7697-37-2	231-714-2	10
Scandium Nitrate	13465-60-6	236-701-5	5.1
Scandium	7440-20-2	231-129-2	1

EC Classification, R/S Phrases: Refer to Section 15, Regulatory Information.

3. HAZARDS IDENTIFICATION

NFPA Ratings (Scale 0-4): Health = 4 Fire = 1 Reactivity = 2

Major Health Hazards: Nitric acid can cause severe or fatal burns if inhaled, swallowed, or absorbed through the skin. Scandium and its compounds may cause skin irritation or other health effects.

Physical Hazards: Containers may break or shatter.

Potential Health Effects

Inhalation:	Nitric acid can damage the mucous membranes and respiratory tract, causing spasm, inflammation of the larynx and bronchi, chemical pneumonitis, and pulmonary edema. Symptoms may include burning sensation, coughing, wheezing, laryngitis, shortness of breath, headache, nausea, and vomiting. Teeth may also be damaged. The toxicity of scandium and its compounds has not been fully investigated, but inhalation of dust or fume may irritate the respiratory tract. Long-term exposure may damage the lungs, liver, or other organs.
Skin Contact:	Nitric acid can cause severe skin burns. Effects of acid burns may be delayed. Contact with scandium and its compounds may cause skin irritation.
Eye Contact:	Nitric acid can cause severe eye irritation, corneal burns, permanent eye damage, or blindness. Contact with scandium and its compounds may cause eye irritation.
Ingestion:	Nitric acid can cause severe burns and damage to the GI tract. Ingestion of scandium compounds may interfere with blood clotting. The toxicity of scandium and its compounds has not been fully investigated, but ingestion may irritate the GI tract. See also Inhalation.

Medical Conditions Aggravated by Exposure: Respiratory tract, skin, eyes, teeth, GI tract, blood.

Listed as a Carcinogen/ Potential Carcinogen:

	Yes	No
In the National Toxicology Program (NTP) Report on Carcinogens	_____	<u> X </u>
In the International Agency for Research on Cancer (IARC) Monographs	_____	<u> X </u>
By the Occupational Safety and Health Administration (OSHA)	_____	<u> X </u>

4. FIRST AID MEASURES

Inhalation: Move the person to fresh air immediately. If not breathing, qualified personnel may start CPR or give oxygen if necessary. Get medical aid at once, and bring the container or label.

Skin Contact: Remove contaminated clothing and shoes. Flush affected skin with water for at least 15 minutes, then wash thoroughly with soap and water. If burns are severe or if skin irritation persists, get medical aid and bring the container or label. Wash contaminated clothing before reusing.

Eye Contact: Remove contact lenses (if any). Do not allow victim to rub eyes or keep eyes closed. Flush eyes with large amounts of running water for at least 30 minutes, keeping eyelids open and raising lids to remove all chemical. Get medical aid at once, and bring the container or label.

Ingestion: Contact a poison control center immediately for instructions. Wash out mouth with water, but do not induce vomiting. Get medical aid at once, and bring the container or label.

Note to Physician (Nitric Acid): Wash affected skin with 5% solution of sodium bicarbonate (NaHCO_2). Activated charcoal is of no value. Do not give bicarbonate to neutralize the material.

5. FIRE FIGHTING MEASURES

Fire and Explosion Hazards: Nitric acid is a powerful oxidizing agent that can react with combustible materials to cause fires. Scandium nitrate is also an oxidizer that may explode on contact with combustible materials. Scandium powder (not present in this SRM) is highly flammable and may ignite or explode. No data are available for the solution, and its behavior may differ from that of the individual components.

Extinguishing Media: Dry chemical, dry salt, or sand. Do not use water or halogenated extinguishers when fighting fires involving scandium. No data are available for the solution, and its behavior may differ from that of the individual components.

Fire Fighting: Avoid inhalation of material or combustion byproducts. Wear full protective clothing and NIOSH-approved self-contained breathing apparatus (SCBA).

Flash Point (°C): N/A

Autoignition (°C): N/A

Lower Explosive Limit (LEL): N/A

Upper Explosive Limit (UEL): N/A

Flammability Class (OSHA): N/A

6. ACCIDENTAL RELEASE MEASURES

Occupational Release: Notify safety personnel of spills. Surfaces contaminated with this material should be covered with soda ash or sodium bicarbonate to neutralize the acid. Place the neutralized material into containers suitable for eventual disposal, reclamation, or destruction.

Disposal: Refer to Section 13, Disposal Considerations.

7. HANDLING AND STORAGE

Storage: Store unopened containers of this material in a dry place with acid-resistant flooring at room temperature. Protect from physical damage, water, humidity, heat, direct sunlight, and incompatible materials.

Safe Handling Precautions: Wear gloves and chemical safety goggles (Section 8). Engineering controls should maintain airborne concentrations below TLV (Section 8).

8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

Nitric Acid:

ACGIH TLV-TWA: 2 ppm or 5 mg/m³

OSHA TLV-TWA: 2 ppm or 5 mg/m³

UK WEL: 5.2 mg/m³

Scandium Nitrate:

OSHA TLV-TWA: None established. Total nuisance dust, 15 mg/m³; respirable dust, 5 mg/m³

ACGIH TLV-TWA: None established. Total nuisance dust, 10 mg/m³; respirable dust, 3 mg/m³

UK WEL: None established. Total inhalable dust, 10 mg/m³; respirable dust, 4 mg/m³

Scandium:

OSHA TLV-TWA: None established. Total nuisance dust, 15 mg/m³; respirable dust, 5 mg/m³

ACGIH TLV-TWA: None established. Total nuisance dust, 10 mg/m³; respirable dust, 3 mg/m³

UK WEL: None established. Total inhalable dust, 10 mg/m³; respirable dust, 4 mg/m³

Ventilation: Use local or general exhaust to keep employee exposures below limits. Local exhaust ventilation is preferred because it can control contaminant emissions at the source, preventing dispersion into the general work area. Refer to the ACGIH document *Industrial Ventilation, a Manual of Recommended Practices*.

Respirator: If necessary, refer to the NIOSH document *Guide to the Selection and Use of Particulate Respirators Certified under 42 CFR 84* for selection and use of respirators certified by NIOSH.

Eye Protection: Use chemical safety goggles where dusting or splashing of solutions may occur. See OSHA standard (29 CFR 1910.133) or European Standard EN166. The employer should provide an emergency eye wash fountain and safety shower in the immediate work area.

Personal Protection: Wear appropriate gloves and protective clothing to prevent contact with skin.

9. PHYSICAL AND CHEMICAL PROPERTIES

Nitric Acid	Scandium Nitrate	Scandium
Appearance and Odor: Colorless to slightly yellow liquid, darkens to brown upon aging and exposure to light; irritating, pungent odor.	Appearance and Odor: Prismatic deliquescent crystals; crystallizes as the tetrahydrate.	Appearance and Odor: Silver or gray metal, odorless; reported to be dimorphic
Relative Molecular Weight: 63.02	Relative Molecular Weight: 230.97 (anhydrous)	Relative Molecular Weight: 44.95
Molecular Formula: HNO ₃	Molecular Formula: Sc(NO ₃) ₃	Molecular Formula: Sc
Specific Gravity: 1.05 (10%)	Specific Gravity: N/A	Specific Gravity: 2.989
Solvent Solubility: Decomposes in alcohol	Solvent Solubility: Soluble in alcohol	Solvent Solubility: Soluble in dilute HCl, releases H ₂ gas
Water Solubility: Soluble	Water Solubility: Soluble	Water Solubility: Soluble, releases H ₂ gas
Boiling Point (°C): 86 (187°F)	Boiling Point: N/A (decomposes)	Boiling Point: 2832 (5130°F)
Vapor Pressure (Pa): 946 @20°C	Vapor Pressure (Pa): N/A	Vapor Pressure (Pa): N/A
Vapor Density (Air=1): 2.17	Vapor Density (Air=1): N/A	Vapor Density (Air=1): N/A
pH: 1.0 (0.1M solution)	pH: N/A	pH: N/A

NOTE: The physical and chemical data provided are for the pure components. Physical and chemical data for this solution do not exist. The actual behavior of the solution may differ from the individual components.

10. STABILITY AND REACTIVITY

Stability: X Stable Unstable

Stable at normal temperatures and pressure.

Conditions to Avoid: Contact with moisture, air, or incompatible materials.

Incompatible Materials:

Nitric Acid: Incompatible with numerous materials including organic materials, plastics, rubber, chlorine, and metal ferrocyanide.

Scandium Nitrate: Incompatible with combustible materials, reducing materials, cyanides, esters, thiocyanates, and sodium hypophosphite.

Scandium: Incompatible with water, oxidizing materials, acids, and halogens.

Fire/Explosion Information: See Section 5.

Hazardous Decomposition: Thermal decomposition of this material may produce nitrogen oxides, scandium oxides, and irritating fumes.

Hazardous Polymerization: _____ Will Occur X Will Not Occur

11. TOXICOLOGICAL INFORMATION

Route of Entry: X Inhalation X Skin X Ingestion

Nitric Acid:

Human, oral: LD_{Lo} = 430 mg/kg

Rat, oral: LD₅₀ > 90 mg/kg

Rat, inhalation: LC₅₀ (4 hrs) = 130 mg/m³

Scandium Nitrate:

No toxicity data found

Scandium:

Human, oral: Preliminary reference dose (RfD) = 16 mg/kg-day

Target Organ(s): Skin, eyes, respiratory tract, GI tract.

Mutagen/Teratogen: Nitric acid has caused birth defects in animals under experimental conditions, and has been investigated as a possible mutagen. The toxicity of scandium and its compounds has not been fully investigated.

Health Effects: See Section 3.

12. ECOLOGICAL INFORMATION

Nitric Acid, Ecotoxicity Data:

Green shore crab (*Carcinus maenas*): LC₅₀ (48 hrs) = 180,000 µg/L

Starfish (*Asterias rubens*): LC₅₀ (48 hrs) = 100,000 to 330,000 µg/L

Brook trout (*Salvelinus fontinalis*): NR-LETH = 1,562 µg/L

Scandium Nitrate: No ecotoxicity data found.

Scandium: No ecotoxicity data found.

Environmental Summary: The ecological effects of this mixture have not been fully evaluated, but one or more components may be slightly toxic to aquatic life. Do not release to the environment.

13. DISPOSAL CONSIDERATIONS

Waste Disposal: One or more components of this mixture is a RCRA hazardous waste. Dispose of container and unused contents in accordance with federal, state, and local requirements for acid waste, which vary according to location. Decontaminate containers before recycling. Processing, use, or contamination of this product may change the waste management options.

14. TRANSPORTATION INFORMATION

U.S. DOT and IATA: Nitric Acid Solution, Hazard Class 8, UN2031, Packing Group II

15. REGULATORY INFORMATION

U.S. REGULATIONS

CERCLA Sections 102a/103 (40 CFR 302.4):

Nitric Acid: RQ = 1000 lb.

Scandium Nitrate: Not regulated

Scandium: Not regulated

SARA Title III Section 302: Nitric acid is regulated

SARA Title III Section 304: Nitric acid is regulated

SARA Title III Section 313: Nitric acid and scandium nitrate (N511, Nitrate Compounds) are regulated.

OSHA Process Safety (29 CFR 1910.119): Nitric acid at higher concentrations ($\geq 94.5\%$) is regulated.

SARA Title III Sections 311/312 Hazardous Categories (40 CFR 370.21):

ACUTE: Yes

CHRONIC: Yes

FIRE: No

REACTIVE: Yes

SUDDEN RELEASE: No

STATE REGULATIONS

California Proposition 65: No components are regulated.

CANADIAN REGULATIONS

WHMIS Classification:

Nitric Acid: C (oxidizing material), D1A (very toxic material), E (corrosive material)

Scandium Nitrate: C (oxidizing material)

Scandium: D2B (toxic material)

WHMIS Ingredient Disclosure List: Nitric acid is regulated.

CEPA Domestic Substances List (DSL): Nitric acid is regulated.

CEPA Non-Domestic Substances List (NDSL): Scandium and scandium nitrate are regulated.

EUROPEAN REGULATIONS

EU/EC Classification:

Nitric Acid: O (Oxidizer), C (Corrosive)

Scandium Nitrate: Xi (Irritant), O (oxidizer); not classified in Annex I of Directive 67/548/EEC.

Scandium: Xi (Irritant); not classified in Annex I of Directive 67/548/EEC.

Risk Phrases (mixture):

R23 (toxic by inhalation)
R25 (toxic if swallowed)
R34 (causes burns)
R36/37/38 (irritating to eyes, respiratory system and skin)

Safety Phrases (mixture):

S20/21 (when using, do not eat, drink or smoke)
S28 (wash after contact with skin)
S45 (in case of accident or illness, see doctor; show label)
S60 (dispose of this material and its container as hazardous waste)

NATIONAL INVENTORY STATUS

U.S. Inventory (TSCA): All components are listed.

TSCA 12(b), Export Notification: No components are listed.

16. OTHER INFORMATION

Sources:

PAN Pesticide Database: Nitric Acid.

Toxicology Excellence for Risk Assessment (TERA), *Development of Reference Doses and Reference Concentrations for Lanthanides*. U.S. Bureau of Land Management, November 1999.

U.S. National Institute for Occupational Safety and Health, *NIOSH Pocket Guide to Chemical Hazards*, September 2005 edition. DHHS (NIOSH) Publication No. 2005-151.

Disclaimer: Physical and chemical data contained in this MSDS are provided only for use as a guide in assessing the hazardous nature of the material. The MSDS was prepared carefully, using current references; however, NIST does not certify the data in the MSDS. The certified values for this material are given in the NIST Certificate of Analysis.